Reactions and enzymes

- 1. Energy is transfered
- 2. Electron is transfered
- 3. Water molecule can be a participant
- 4. Metabolic pathways
- 5. Enzymes

Energy is transfered

- Exergonic reaction: energy-generating reaction (reaction gives energy)
- Endergonic reaction: energy-consuming reaction (reaction uses energy)

Electron is transfered

- Oxidation: reactant (A) loses electrons, reactant is oxidized
- Reduction: reactant (B) gains electrons, reactants is reduced
- Oxidant= oxidizing agent (B): reactant gaining electron
- Reductant (A): reactant losing electron



Water molecule can be a participant

- condensation (dehydration): formation of a bond between 2 reactants accompanied by formation of a H₂O molecule, eg. peptide bond formation, ester bond formation
- hydrolysis (hydration): breakdown of a bond accompanied by breakdown of a H₂O molecule, eg. breakdown of peptide bond, breakdown of ester bond



Metabolic pathways

- a series of individual chemical reactions in a living system
- product of one reaction in a pathway serves as the reactant for the following reaction
- always accompanied by energy transfer and electron transfer (electron transporters are involved eg. NAD)



Metabolic pathways

- <u>anabolic pathway</u>= biosynthetic pathway: synthesis (production) of a molecule
 - needs energy (eg. energy of sunlight or ATP)
 - reactants are reduced
 - example:

photosynthesis: synthesis of glucose from CO₂ and water in plants (needs energy of sunlight)

- <u>catabolic pathway:</u> breakdown of a molecule
 - produces energy (ATP, heat)
 - reactants are oxidized
 - examples:
 - biological oxidation of glucose (aerobic cellular respiration): breakdown of glucose into CO₂ and water
 - Fermentation (anaerobic cellular respiration): breakdown of glucose into lactate and CO₂



 $1 C_6 H_{12} O_6 = 6 CO_2 + 6 H_2 O + 36 ATP$

Glycolysis

Krebs cycle/citrate cycle



Panel 13-1 Details of the 10 steps of glycolysis







www.uic.edu/classes/bios/bios100/summer2003/krebsfull.htm



- reaction is faster at the presence of an enzyme (even 10 000 times faster)
- reaction needs less activation energy at the presence of an enzyme (even 10 times less)



Mechanism of enzyme activity



- 1. reactant (=substrate) binds to the active center (or site) of the enzyme (<u>..lock and key" model</u>)
- 2. chemical reaction is performed (substrate is chemically modified) in enzyme-substrate complex: product is made
- 3. product leaves enzyme
- 4. enzyme can bind substrates again

Hydrolysis of sucrose



- 1. chemically <u>proteins (some enzymes are chemically RNA, eg. peptidyl transferase)</u>
- 2. Holoenzyme (eg.holoenzyme for DNA synthesis=replisome)
- 3. Coenzymes (eg. coenzyme A), electron transporters (eg. NAD, FAD)
- 4. sensitivity for pH and temperature



- 5. Scientific name: name of substrate+name of reaction+ASE (eg. Glycogensynthase)
- 6. examples:
- digestive enzymes: catalyze breakdown of nutrients in digestive system eg.: amylase (breakdown of carbohydrates) pepsin (breakdown of proteins), lipase(breakdown of lipids)
- biosynthetic enzymes: DNA polymerase (synthesis of DNA), RNA polymerase (synthesis of RNA), peptidyl transferase (synthesis of proteins)

replisome



Albinism=hypomelanism





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